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11.1 Describing Chemical Reactions 11
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2. Sulfur burns in oxygen to form sulfur dioxide. Write a skeleton equation for this chemical reaction. Analyze Step 1. Write the formula for each reactant and each product. Include the common STP state of each substance.

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Chapter 11 Chemical Reactions 119

GUIDED PRACTICE PROBLEM 2 (page 324) 2. Sulfur burns in oxygen to form sulfur dioxide. Write a skeleton equation for this chemical reaction. Include appropriate symbols from table 11.1. Analyze Step 1. Identify the relevant concepts. Write the

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formula for each reactant and each product.
Include the common

SECTION 11.1 DESCRIBING
CHEMICAL REACTIONS (pages
321 – 329)

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Neutralization Reactions, Part 1 Chapter 11

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Study ... Chemistry is the study of how

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Reactions 119 GUIDED PRACTICE

PROBLEM 2 (page 324) 2. Sulfur burns in oxygen to form sulfur dioxide. Write a skeleton equation for this chemical reaction.

Include appropriate symbols from table 11.1. Analyze Step 1. Identify the relevant

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guided tutorial. Decomposition Reactions

When mercury(II) oxide is heated, it decomposes or breaks down into two simpler compounds, as shown in Figure 11.6. A

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decomposition reaction is a chemical change in which a single compound breaks down into two or more simpler products.

Decomposition reactions involve only one reactant and two or more ...

11.2 Types of Chemical Reactions

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Balancing Equations: Answers to Practice

Problems 1. Balanced equations.

(Coefficients equal to one (1) do not need to be shown in your answers).

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