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Chapter 8 Covalent Bonding 183 Section Review Objectives • State a rule that usually tells how many electrons are shared to form a covalent bond • Describe how electron dot formulas are used • Predict when two atoms are likely to be joined by a double or a triple covalent bond • Distinguish between a single covalent bond and other covalent bonds

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化学键the force that holds two atoms together. 共价键the chemical bond that results from sharing valence electro.... 分子a molecule is formed when two or more atoms bond covalently. 电子点式used to show valence electrons of atoms.

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page 244 Draw the Lewis structure for each molecule 1 PH₃ H₂ H₂H—H₂H₂P respectively, for single, double, and triple P—
— 2 H₂S H₂H₂H—H₂S

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